

Scintillation properties deterioration due to hydroxyl and oxygen presence in Eu doped alkali halides

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Introduction

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Materials and methods

CsI:Eu crystals were grown by Czochralski method. Spectral characteristics of photoemission were studied using FLS920 fluorescence spectrometer. IR-absorption spectra were recorded by Spectrum One FT-IR spectrometer.

X-ray luminescence spectra were measured by X-ray tube (Wanode, 30 mA, 30 kV). Irradiation of the samples was realized by X-ray tube (W anode, 10 mA, 150 kV). Measurement of thermostimulation luminescent curves was carried out at heating rate of 0.08 K/sec and 0.2 K/sec in the range of 20-350 K and 290-550 K correspondingly.

It is known that oxygen traces can significantly change radiation stability, emission spectra, light yield and afterglow level in conventional scintillators [1]. The last trend to development of hygroscopic Eu-doped alkali- and alkali-earth halides claims for an extra pure raw material and dry or vacuum atmosphere during crystal growth and processing [2].

In general the study of oxygen-containing impurity influence on luminescence is an actual problem for conventional and new halide scintillator.

Luminescence study of Eu²⁺ doped CsI crystals deals mainly with the afterglow suppression in CsI(TI) scintillators [3-6]. Afterglow is usually associated with oxygencontaining impurities. At the same time it was proved that Eu²⁺ ions react with oxygen ions and act as a scavenger in growing process. Along with that, a part of europium ions penetrates into CsI crystal which leads to the formation of emission centers [5].

The entering of Eu²⁺ ions into the CsI lattice requires charge compensation. It may be performed by cation vacancy and/or anion impurity. This was the reason for choosing CsI:Eu crystals as suitable object for Eu^{2+} emission details study.

The aim of the work is to study the luminescence properties deterioration due to hydroxyl / oxygen presence in Eu doped CsI crystals.

The nature of emission centers in CsI:Eu

Photo luminescence CsI:Eu (10⁻³ mass%)



- **Luminescence spectra depend on Eu concentration and excitation region.**
- C1 center is caused by the radiative relaxation at the vacancy type defect, localized near the Eulion.
- \equiv C2 center is typical for d-f transitions in Eu²⁺.
- \times C3 center dominates at high concentration (up to 10⁻¹ mass%).

Results

1.0

08

0.6

0.4

0.2

3.6

3,4

3,2

3,0

2,8

E, eV

a.u.

Exc 340nm

Heat treatment and radiation influence

Quenching and Annealing

Emission centers in CsI:Eu are stable up to T \approx 300°C. Restructuring or aggregation of centers was not found in treatment process. This fact contradicts the previous concepts on the decisive role of Eu²⁺-u_c⁻ dipoles aggregation in process of storage and low temperature annealing of crystals CsI:Eu [7, 8]. The rise of annealing temperature up to 450°C followed by quenching leads to increase of the C3 (460 nm) centers and reduce of C1 (410 nm), C2 (440 nm) and C4 (480-530 nm) bands.



X-ray luminescence

λ, **nm**

Quenched

2,6 2,4 2,2

Initial







IR absorption as an evidance of oxygen containing anions presence



IR bands of H₂O, OH⁻ were revealed. The total content of oxygen radicals is comparable with the Eu^{2+} concentration in the crystal (10⁻³ mass%). Therefore, it can be assumed that the hydroxyl / oxygen ions may be a part of luminescence centers.

Luminescence centers in CsI:Eu

Center	Photo emission, nm	Supposed nature
<i>C1</i>	410	Eu ²⁺ near defect
<i>C2</i>	442	Eu ²⁺ (symmetric)
<i>C3</i>	455	$Eu^{2+}(H_2O^-)v_c^-$
<i>C4</i>	480-530	$O^{2-} / Eu^{2+} - O^{2-}$

Dose dependences of the luminescence yield of CsI:Eu



- x Irradiation leads to the decrease of the integral light output and change of luminescence spectra.
- π At the initial stages of irradiation (D < 50 Gy) the most notable is the suppression of C1, C3, C4 centers.
- π The higher doses cause the decrease of C1 center.
- Process is accompanied by the TSL peaks emergence and growth, which indicates the formation of intrinsic and activator color centers.

Conclusions

Several types of activator centers were revealed in CsI:Eu crystals. Emission centers are stable at room temperature but their structure and concentration change under heat treatment and irradiation.

The irradiation leads to decrease of emission efficiency related with the charge carriers capturing and restructuring of activator centers. The nature of some emission centers is determined by the presence of intrinsic and extrinsic defects near Eu^{2+} ions. The dominant luminescent centers include not only vacancies, but also oxygen and/or hydroxylions.

Reference:

1. "Inorg. Scintillators for Detector Systems" Ed. P. Lecoq, A. Annenkov, A. Gektin, M. Korzhik, C. Pedrini, Springer, 2006, 287 p.

2. A. Gektin, N. Shiran, "Scintillation Materials: engineering, devices and application" Kharkov, ISMA, (2011) 6-39.

3. L.A. Kappers, R.H. Bartram, D.S. Hamilton et al. Rad. Meas., 42 (2007) 537.

4. M. Kudin, A. I. Mitichkin, T. A. Charkina et. al. J. Opt. Techn. 74 (2007) 633.

5 N. Shiran, A. Gektin, S. Vasyukov et. al. Funct. Mater. 18 (2011) 438.

6. A. Gektin, N. Shiran, A. Belsky, S. Vasyukov, Opt. Mater. 34 (2012) 2017.

7. V. Savel'ev, V. Avdonin, L. Dugarova, et al., Sov. Phys. Sol. St. 16 (1974) 1090.

8. V. Yakovlev, L. Trefilova, A. Meleshko, N. Ovcharenko, J. Lumin.132 (2012) 2476.

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